

34405-42-0; 4, 34405-43-1; 5, 122-09-8; 5 HCl, 1197-21-3; 6, 5531-33-9.

Acknowledgment.—We wish to acknowledge the support of Dr. J. B. Ziegler and helpful discussions with Mr. L. Dorfman, whose staff we thank for microanalyses and spectra.

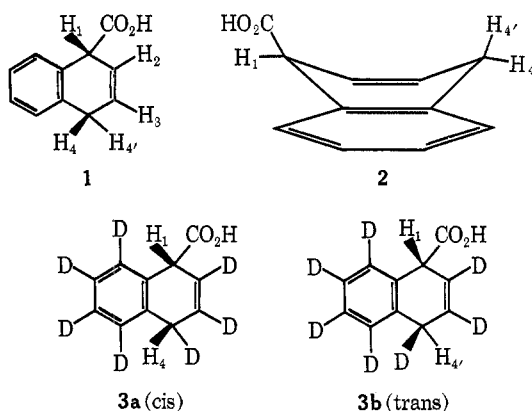
The Conformation of 1,4-Dihydro-1-naphthoic Acid. II. The Nuclear Magnetic Resonance Spectrum of the Heptadeuterio Analog

J. L. MARSHALL,* A. M. IHRIG, AND P. N. JENKINS

Department of Chemistry, North Texas State University,
Denton, Texas 76203

Received October 19, 1971

We previously reported¹ that the pmr spectrum of 1,4-dihydro-1-naphthoic acid (**1**) argued for a puckered conformation of the dihydro ring with the carboxylate group in the pseudoaxial position (see **2**) and that all nmr parameters of **1** could be determined from this study except the homoallylic coupling constants J_{14} and $J_{14'}$. We now wish to report the determination of these homoallylic parameters from the pmr spectrum of the heptadeuterio analog **3** and to present a more complete analysis of the conformation of **1**.



The heptadeuterio compound **3** was synthesized in a three-step sequence from perdeuterionaphthalene (see Experimental Section). The deuterium-decoupled pmr spectrum of **3** demonstrated an approximate 50:50 mixture of the *cis*- and *trans*-dihydro epimers (**3a** and **3b**) from the approximately equal areas corresponding to H_4 and $H_{4'}$. The H_4 and $H_{4'}$ signals were split into doublets, directly giving $J_{14} = J_{cis} = 3.84$ Hz and $J_{14'} = J_{trans} = 4.36$ Hz.² Since for the proposed conformation of **1** it is to be expected³ that $J_{14} > J_{14'}$ these newly determined parameters are consistent with our previous contention that the carboxylate group was pseudoaxial.¹

This completion of the determination of the pmr parameters for **1** allows a fuller analysis of the conformation of **1**. The similar values of J_{14} and $J_{14'}$ strongly

suggest the dihydro ring is nearly flat.⁴ Furthermore, a closer inspection of the previously determined parameters¹ of **1** also indicates the dihydro ring is not strongly puckered. First, if the dihydro ring of **1** were a true boat, the dihedral angle involving H_3 and $H_{4'}$ would be near 90° and $J_{34'}$ should be much less than the observed value of 2.44 Hz.⁶ Second, the allylic coupling constants allow the calculation⁵ that the dihedral angle involving H_2 and H_4 is much larger than 0° .⁷ Third, the absolute value of $J_{44'}$ ¹ is suspiciously large for a highly puckered system,⁸ and is much more consistent with a nearly planar ring. Thus, it appears that the conformation of **1** is a "flattened boat" in which the dihydro ring is only slightly puckered and that the pmr data for **1** lead to conclusions consistent with work for other 1,4-cyclohexadienes, in which it has been proposed that this system is planar or only slightly puckered.⁹

Experimental Section

Nmr spectra were recorded on a JEOL PS-100 spectrometer, using tetramethylsilane as the internal standard and deuteriochloroform as the solvent. Deuterium-decoupling was done with a JEOL deuterium radiofrequency oscillator JNM-RH-D, in conjunction with a JEOL heteronuclear decoupler JNM-SD-HC. Melting points were determined by a Thomas-Hoover melting point apparatus. All deuterated compounds were analyzed by pmr and were found to have a minimum isotopic purity of 98%.

Naphthalene- d_8 (minimum isotopic purity of 98%) was purchased from Diaprep, Inc., Atlanta, Ga.

1-Bromonaphthalene- d_7 was synthesized from naphthalene- d_8 in the fully developed bromination procedure¹⁰ to give a 62% yield (9.78 g), bp $95-112^\circ$ (0.5 mm) [lit.¹⁰ bp $132-135^\circ$ (12 mm), 1-bromonaphthalene].

1-Naphthoic acid- d_7 was synthesized from the bromo precursor by the usual Grignard procedure¹¹ to give an 88% yield (7.37 g), mp (H_2O) $155-159^\circ$ (lit.¹² mp $159-160^\circ$, 1-naphthoic acid).

2,3,4,5,6,7,8-Heptadeuterio-1,4-dihydro-1-naphthoic acid (3**)** was prepared by the Birch reduction of the perdeuterionaphthoic precursor in the previous manner¹ to give a 73% yield (1.79 g), mp (hexane) $86-88^\circ$ (lit.¹³ mp 86° , **1**).

Registry No.—**1**, 5111-73-9; **3a**, 34405-19-1; **3b**, 34405-20-4.

Acknowledgment.—Acknowledgment is made to the Robert A. Welch Foundation (Grant No. B-325) and to North Texas State University for a Faculty Research Grant for support of this work. In addition, appreciation is extended to Dr. S. Sternhell and M. Barfield for stimulating discussions and suggestions for analysis of the data.

(4) The Barfield INDO treatment⁵ predicts that, for a flat dihydro ring, $J_{14'}/J_{14} = 1.12$ and that this ratio increases with increased puckering to 3.3 for a true boat (S. Sternhell, private communication). The determined ratio is actually 1.14.

(5) M. Barfield, *J. Amer. Chem. Soc.*, **93**, 1066 (1971).

(6) K. L. Williamson and W. S. Johnson, *ibid.*, **83**, 4623 (1961); M. Karplus, *J. Chem. Phys.*, **30**, 11 (1959); S. Sternhell, *Quart. Rev., Chem. Soc.*, **23**, 236 (1969); A. A. Bothner-By, S. Castellano, S. J. Ebersole, and H. Gunther, *J. Amer. Chem. Soc.*, **88**, 2466 (1966).

(7) It can be predicted⁵ from the pmr data¹ of 1,4-dihydro-1-naphthoic acid that the dihedral angle involving H_2 and H_4 is $\sim 52^\circ$ and that involving H_2 and $H_{4'}$ is $\sim 68^\circ$.

(8) M. Barfield and D. M. Grant, *J. Amer. Chem. Soc.*, **85**, 1899 (1963).

(9) H. Oberhammer and S. H. Bauer, *ibid.*, **91**, 10 (1969); M. J. Bennett, J. T. Purdham, S. Takada, and S. Masamune, *ibid.*, **93**, 4063 (1971); R. J. Jandacek and S. H. Simonsen, *ibid.*, **91**, 6663 (1969).

(10) H. T. Clarke and M. R. Brethen, "Organic Syntheses," Collect. Vol. I, Wiley, New York, N. Y., 1932, p 121.

(11) J. L. Marshall, K. C. Erickson, and T. K. Folsom, *J. Org. Chem.*, **35**, 2038 (1970).

(12) A. A. Morton, J. B. Davidson, T. R. P. Gibb, Jr., E. L. Little, E. F. Clarke, and A. G. Green, *J. Amer. Chem. Soc.*, **64**, 2250 (1942).

(13) K. Von Auwers and K. Moller, *J. Prakt. Chem.*, **109**, 144 (1925).

(1) J. L. Marshall and T. K. Folsom, *J. Org. Chem.*, **36**, 2011 (1971).

(2) The methine signal (H_1) was also split into a doublet with $J_{obsd} = 4.1$ Hz, but resolution was not sufficient to distinguish the two J values.

(3) M. Karplus, *J. Chem. Phys.*, **33**, 1842 (1960); M. Barfield, *ibid.*, **48**, 4463 (1968).